Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

(*E*)-3-[4-(Dimethylamino)phenyl]-1-(2-pyridyl)prop-2-en-1-one

Songzhu Lin,* Ruokun Jia, Xiaojun Zhang, Zhiwen Wang and Yanlin Yuan

College of Chemical Engineering, Northest Dianli University, 132012 Jilin, People's Republic of China
Correspondence e-mail: Songzhulin@hotmail.com

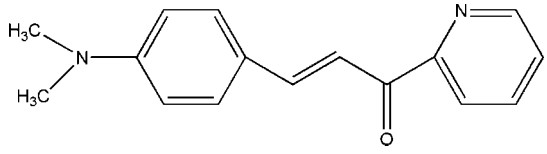
Received 7 April 2009; accepted 23 April 2009

Key indicators: single-crystal X-ray study; T = 295 K; mean σ(C–C) = 0.003 Å; R factor = 0.044; wR factor = 0.130; data-to-parameter ratio = 15.0.

In the title molecule, $C_{16}H_{16}N_2O$, the pyridine ring and non-H atoms of the ═CH—C(═O)— unit are coplaner, the largest deviation being 0.045 (2) Å for the O atom. The dihedral angle between this plane and the benzene ring is 2.79 (2)°. The molecular structure is stabilized by intermolecular C—H$\cdots\pi$ and interactions.

Related literature

For a related structure, see: Butcher *et al.* (2007). For the phamacological activity of chalcones, see: Zhao *et al.* (2007); Fichou *et al.* (1988). For the blue-light transmittance of chalcone derivatives, see: Sarojini *et al.* (2006).



Experimental

Crystal data

$C_{16}H_{16}N_2O$
M_r = 252.31
Monoclinic, $P2_1/c$
a = 8.1553 (4) Å
b = 17.4543 (12) Å
c = 12.1087 (5) Å
β = 125.032 (5)°
V = 1411.35 (16) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.08 mm^{-1}
T = 295 K
0.25 × 0.20 × 0.18 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: none
7572 measured reflections
2619 independent reflections
1608 reflections with $I > 2\sigma(I)$
R_{int} = 0.026

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.044
$wR(F^2)$ = 0.130
S = 1.02
2619 reflections
175 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.13 e Å^{-3}
$\Delta\rho_{min}$ = −0.10 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H$\cdots A$	D—H	H$\cdots A$	$D\cdots A$	D—H$\cdots A$
C10—H10$A\cdots Cg1^{i}$	0.93	2.90	3.662	139
C15—H15$B\cdots Cg2^{ii}$	0.96	3.20	3.870	128
C16—H16$B\cdots Cg1^{iii}$	0.96	3.17	3.908	135

Symmetry codes: (i) $x, -y-\frac{1}{2}, z-\frac{3}{2}$; (ii) $-x+1, -y, -z$; (iii) $x-1, y, z-1$. *Cg*1 and *Cg*2 are the centroids of the pyridine and phenyl rings, respectively.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2499).

References

Butcher, R. J., Jasinski, J. P., Narayana, B., Mayekar, A. N. & Yathirajan, H. S. (2007). *Acta Cryst.* E**63**, o3546.
Fichou, D., Watanabe, T., Takeda, T., Miyata, S., Goto, Y. & Nakayama, M. (1988). *Jpn J. Appl. Phys.* **27**, 429–430.
Sarojini, B. K., Narayana, B., Ashalatha, B. V., Indira, J. & Lobo, K. J. (2006). *J. Cryst. Growth*, **295**, 54–59.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Zhao, P. L., Liu, C. L., Huang, W., Wang, Y. Z. & Yang, G. F. (2007). *J. Agric. Food Chem.* **55**, 5697–5700.

supplementary materials

Acta Cryst. (2009). E**65**, o1161 [**doi:10.1107/S1600536809015244**]

(*E*)-3-[4-(Dimethylamino)phenyl]-1-(2-pyridyl)prop-2-en-1-one

S. Lin, R. Jia, X. Zhang, Z. Wang and Y. Yuan

Comment

As an intermediate in the biosynthetic pathway of flavonoids, isoflavonoids, and aurone, chalcones have been shown to display a diverse array of phamacological activities, among which are antifungal, antibacterial, antiprotozoal, anti-inflammatory, antitumor, antimalarial, and anti-HIV activities (Zhao *et al.*, 2007; Fichou *et al.*, 1988; Butcher *et al.*, 2007). In addition, chalcone derivatives are noticeable materials for their excellent blue light transmittance and good crystallizability (Sarojini *et al.*, 2006). In order to research this kind of complex, we synthesis the title compound (I) and report its crystal structure (Fig. 1).

In the title molecule, $C_{16}H_{16}N_2O$, the pyridine ring and the atoms C6,C7,O1 are coplaner (p1), with the largest deviation of 0.045Å for O1. The dihedral angle between p1 and phenyl ring is 2.79 (2)°.

It is interesting to note that the molecular structure is stabilized by intermolecular C—H···π interactions and C—H···N intramolecular interactions (Table 1) [C_g(1) and C_g(2) refer to pyridine and phenyl ring, respectively].

Experimental

5 ml of 10% KOH solution was added to solution of 2-acetylpyridine (1.21 g, 0.01 mol) and 4-(dimethylamino)benzaldehyde (1.49 g, 0.01 mol) in 30 ml ethanol. The solution was stirred for 10 h and filtered. The product obtained was crystallized from acetone/toluene (1:1).

Refinement

All H atoms were placed in calculated positions, with C—H=0.93–0.96 Å, and included in the final cycles of refinement using a riding model, with U_{iso}(H)=1.2–1.5 times U_{eq}(C).

Figures

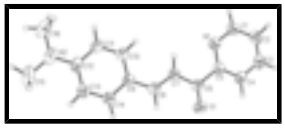


Fig. 1. The molecular structure of (I), with atom labels and 30% probability displacement ellipsoids for non-H atoms.

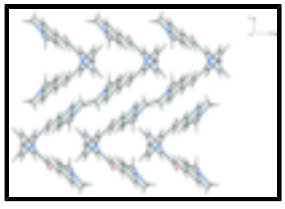


Fig. 2. The packing of (I), viewed down the *c* axis.

(*E*)-3-[4-(Dimethylamino)phenyl]-1-(2-pyridyl)prop-2-en-1-one

Crystal data

$C_{16}H_{16}N_2O$	F_{000} = 536
M_r = 252.31	D_x = 1.187 Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation λ = 0.71073 Å
Hall symbol: -p 2y bc	Cell parameters from 1586 reflections
a = 8.1553 (4) Å	θ = 2.0–25.4º
b = 17.4543 (12) Å	μ = 0.08 mm^{-1}
c = 12.1087 (5) Å	T = 295 K
β = 125.032 (5)º	Block, red
V = 1411.35 (16) Å^3	0.25 × 0.20 × 0.18 mm
Z = 4	

Data collection

Bruker SMART CCD area-detector diffractometer	1608 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	R_{int} = 0.026
Monochromator: graphite	θ_{max} = 25.5º
T = 295 K	θ_{min} = 2.3º
φ and ω scans	h = −6→9
Absorption correction: none	k = −21→20
7572 measured reflections	l = −14→14
2619 independent reflections	

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)]$ = 0.044	$w = 1/[\sigma^2(F_o^2) + (0.054P)^2 + 0.1706P]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2)$ = 0.130	$(\Delta/\sigma)_{max} < 0.001$
S = 1.02	$\Delta\rho_{max}$ = 0.13 e Å^{-3}
2619 reflections	$\Delta\rho_{min}$ = −0.10 e Å^{-3}
175 parameters	Extinction correction: SHELXL97 (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.025 (3)
Secondary atom site location: difference Fourier map	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.3917 (2)	0.59831 (8)	−0.12014 (13)	0.0974 (5)
N1	0.1005 (2)	0.75477 (9)	−0.13939 (15)	0.0741 (4)
N2	0.7635 (3)	0.56873 (11)	0.70210 (17)	0.0942 (6)
C1	−0.0344 (3)	0.80574 (12)	−0.2313 (2)	0.0888 (6)
H1A	−0.0832	0.8425	−0.2015	0.107*
C2	−0.1059 (3)	0.80743 (13)	−0.3672 (2)	0.0873 (6)
H2A	−0.1993	0.8438	−0.4256	0.105*
C3	−0.0344 (3)	0.75393 (13)	−0.41174 (19)	0.0831 (6)
H3A	−0.0778	0.7532	−0.5015	0.100*
C4	0.1047 (3)	0.70057 (11)	−0.32047 (18)	0.0717 (5)
H4A	0.1552	0.6637	−0.3489	0.086*
C5	0.1685 (2)	0.70242 (9)	−0.18518 (17)	0.0606 (4)
C6	0.3188 (3)	0.64457 (10)	−0.08317 (17)	0.0665 (5)
C7	0.3733 (3)	0.64595 (10)	0.05562 (17)	0.0662 (5)
H7A	0.3199	0.6844	0.0793	0.079*
C8	0.4975 (3)	0.59389 (9)	0.15074 (17)	0.0644 (5)
H8A	0.5468	0.5563	0.1227	0.077*
C9	0.5646 (2)	0.58887 (9)	0.29150 (16)	0.0598 (4)
C10	0.4985 (3)	0.63890 (10)	0.34847 (18)	0.0723 (5)
H10A	0.4098	0.6779	0.2955	0.087*
C11	0.5612 (3)	0.63226 (11)	0.48151 (19)	0.0784 (6)
H11A	0.5121	0.6666	0.5146	0.094*
C12	0.6978 (3)	0.57483 (11)	0.56871 (18)	0.0704 (5)
C13	0.7654 (3)	0.52452 (11)	0.51256 (18)	0.0728 (5)
H13A	0.8554	0.4859	0.5656	0.087*
C14	0.6996 (3)	0.53165 (10)	0.37868 (18)	0.0680 (5)
H14A	0.7470	0.4970	0.3449	0.082*
C15	0.9044 (4)	0.50919 (15)	0.7917 (2)	0.1197 (9)
H15A	1.0236	0.5121	0.7937	0.180*
H15B	0.8437	0.4597	0.7591	0.180*
H15C	0.9381	0.5167	0.8811	0.180*
C16	0.6813 (4)	0.61785 (15)	0.7560 (2)	0.1247 (10)
H16A	0.5384	0.6120	0.7031	0.187*

H16B	0.7140	0.6703	0.7528	0.187*
H16C	0.7371	0.6037	0.8478	0.187*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.1348 (12)	0.0967 (10)	0.0907 (10)	0.0326 (9)	0.0822 (10)	0.0081 (8)
N1	0.0852 (11)	0.0825 (10)	0.0739 (10)	0.0124 (9)	0.0569 (9)	0.0096 (8)
N2	0.1095 (14)	0.1125 (14)	0.0666 (11)	−0.0230 (11)	0.0540 (10)	−0.0111 (10)
C1	0.1010 (15)	0.0973 (15)	0.0916 (14)	0.0264 (13)	0.0690 (13)	0.0204 (12)
C2	0.0818 (14)	0.1094 (16)	0.0779 (13)	0.0158 (12)	0.0500 (12)	0.0254 (12)
C3	0.0809 (13)	0.1099 (16)	0.0624 (11)	−0.0060 (12)	0.0433 (11)	0.0071 (11)
C4	0.0804 (13)	0.0803 (12)	0.0680 (12)	−0.0099 (10)	0.0506 (10)	−0.0070 (10)
C5	0.0655 (10)	0.0652 (10)	0.0655 (10)	−0.0091 (9)	0.0459 (9)	−0.0026 (8)
C6	0.0782 (12)	0.0663 (10)	0.0724 (11)	−0.0016 (9)	0.0534 (10)	−0.0032 (9)
C7	0.0756 (11)	0.0681 (11)	0.0684 (11)	0.0040 (9)	0.0491 (10)	−0.0014 (9)
C8	0.0711 (11)	0.0635 (10)	0.0718 (11)	−0.0008 (9)	0.0488 (10)	−0.0052 (9)
C9	0.0618 (10)	0.0621 (10)	0.0624 (10)	−0.0032 (8)	0.0398 (9)	−0.0043 (8)
C10	0.0753 (12)	0.0755 (12)	0.0694 (12)	0.0078 (9)	0.0434 (10)	−0.0048 (9)
C11	0.0841 (13)	0.0875 (13)	0.0743 (13)	−0.0014 (11)	0.0517 (11)	−0.0168 (11)
C12	0.0712 (12)	0.0834 (13)	0.0610 (11)	−0.0225 (10)	0.0404 (10)	−0.0124 (9)
C13	0.0752 (12)	0.0774 (12)	0.0690 (12)	0.0005 (10)	0.0431 (10)	0.0051 (9)
C14	0.0734 (12)	0.0678 (11)	0.0743 (12)	0.0026 (9)	0.0492 (10)	−0.0012 (9)
C15	0.122 (2)	0.150 (2)	0.0671 (14)	−0.0307 (18)	0.0424 (14)	0.0099 (15)
C16	0.170 (2)	0.141 (2)	0.0992 (17)	−0.0506 (19)	0.0988 (18)	−0.0452 (16)

Geometric parameters (Å, °)

O1—C6	1.2293 (19)	C8—C9	1.463 (2)
N1—C5	1.343 (2)	C8—H8A	0.9300
N1—C1	1.355 (2)	C9—C10	1.398 (2)
N2—C12	1.383 (2)	C9—C14	1.411 (2)
N2—C16	1.454 (3)	C10—C11	1.387 (2)
N2—C15	1.465 (3)	C10—H10A	0.9300
C1—C2	1.396 (3)	C11—C12	1.418 (3)
C1—H1A	0.9300	C11—H11A	0.9300
C2—C3	1.365 (3)	C12—C13	1.403 (2)
C2—H2A	0.9300	C13—C14	1.390 (2)
C3—C4	1.391 (3)	C13—H13A	0.9300
C3—H3A	0.9300	C14—H14A	0.9300
C4—C5	1.406 (2)	C15—H15A	0.9600
C4—H4A	0.9300	C15—H15B	0.9600
C5—C6	1.520 (2)	C15—H15C	0.9600
C6—C7	1.472 (2)	C16—H16A	0.9600
C7—C8	1.356 (2)	C16—H16B	0.9600
C7—H7A	0.9300	C16—H16C	0.9600
C5—N1—C1	116.17 (16)	C10—C9—C8	123.02 (16)
C12—N2—C16	120.7 (2)	C14—C9—C8	121.56 (15)

C12—N2—C15	122.19 (19)	C11—C10—C9	122.09 (17)
C16—N2—C15	116.96 (19)	C11—C10—H10A	119.0
N1—C1—C2	125.10 (19)	C9—C10—H10A	119.0
N1—C1—H1A	117.4	C10—C11—C12	122.16 (17)
C2—C1—H1A	117.4	C10—C11—H11A	118.9
C3—C2—C1	117.86 (19)	C12—C11—H11A	118.9
C3—C2—H2A	121.1	N2—C12—C13	121.42 (19)
C1—C2—H2A	121.1	N2—C12—C11	122.41 (18)
C2—C3—C4	118.83 (18)	C13—C12—C11	116.17 (16)
C2—C3—H3A	120.6	C14—C13—C12	120.88 (18)
C4—C3—H3A	120.6	C14—C13—H13A	119.6
C3—C4—C5	119.88 (17)	C12—C13—H13A	119.6
C3—C4—H4A	120.1	C13—C14—C9	123.30 (16)
C5—C4—H4A	120.1	C13—C14—H14A	118.4
N1—C5—C4	122.16 (17)	C9—C14—H14A	118.4
N1—C5—C6	116.63 (15)	N2—C15—H15A	109.5
C4—C5—C6	121.22 (15)	N2—C15—H15B	109.5
O1—C6—C7	122.33 (17)	H15A—C15—H15B	109.5
O1—C6—C5	118.26 (15)	N2—C15—H15C	109.5
C7—C6—C5	119.41 (15)	H15A—C15—H15C	109.5
C8—C7—C6	123.19 (16)	H15B—C15—H15C	109.5
C8—C7—H7A	118.4	N2—C16—H16A	109.5
C6—C7—H7A	118.4	N2—C16—H16B	109.5
C7—C8—C9	128.81 (16)	H16A—C16—H16B	109.5
C7—C8—H8A	115.6	N2—C16—H16C	109.5
C9—C8—H8A	115.6	H16A—C16—H16C	109.5
C10—C9—C14	115.41 (15)	H16B—C16—H16C	109.5

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C7—H7A···N1	0.93	2.51	2.846 (2)	102
C8—H8A···O1	0.93	2.55	2.873 (2)	101
C10—H10A···Cg1^{i}	0.93	2.90	3.662	140
C15—H15B···Cg2^{ii}	0.96	3.20	3.870	128
C16—H16B···Cg1^{iii}	0.96	3.17	3.908	135

Symmetry codes: (i) *x*, −*y*−1/2, *z*−3/2; (ii) −*x*+1, −*y*, −*z*; (iii) *x*−1, *y*, *z*−1.

Fig. 1

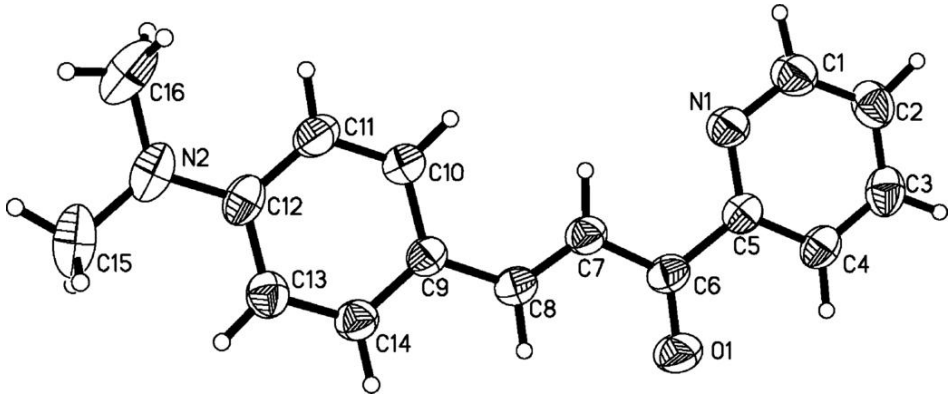


Fig. 2